## B.E. Chemical (4<sup>th</sup> year- 7<sup>th</sup> Sem)\_Environmental Engineering\_20<sup>th</sup> Jan, 2022

## MID SEMESTOR EXAM-II

Attempt all questions: Marks:25

## ASSUME ANY MISSING DATA

- 1. A completely mixed activated sludge process is to be used to treat wastewater flow of 750 m<sup>3</sup>/hr having a soluble BOD<sub>5</sub> of 280 mg/l. The conc. of soluble BOD<sub>5</sub> escaping the treatment is 25 mg/l. The saturation constant =110 mg/l, endogeneous decay coeff is 0.07 day<sup>-1</sup>, yield coeff is 0.6, max specific substrate utilization rate; k=5 day<sup>-1</sup> and the conc of MLVSS is 2100 mg/l. Density of air at  $25^{\circ}$ C= 1.185 kg/m<sup>3</sup>; Calculate:
- a) the mean cell residence time ----(1)
- b) the hydraulic retention time .....(1)
- c) the treatment efficiency and F/M ratio ......(1)
- d) the volume of aeration tank.....(1)
- e)  $P_x$  and  $mO_2$  (in kg/day).....(3)

If air is supplied to the plant at  $25^{\circ}$ C and the oxygen transfer efficiency is 15%, BOD<sub>5</sub> is 72.5% of the ultimate BOD, calculate the volume of air supplied to the plant in m<sup>3</sup>/day. ...(3)

**CO5** 

- 2. Explain the following in detail:
- a) Types of solid wastes
- b) Collection and transportation of MSW
- c) Sanitary landfill and incineration
- d) Pyrolysis and gasification

**CO6** (8)

3. a) A town discharges 19,000 m<sup>3</sup>/day of sewage into a nearby stream. The stream has a minimum flow of 0.5 m<sup>3</sup>/s, depth 3 m and velocity 6 kmph. The data available is:

	TEMP <sup>0</sup> C	DO (mg/l)	$BOD_5 (mg/l)$
STREAM:	20	8.5	10
SEWAGE:	25	1.0	200

The deoxygenating constant ( $k_1$ ) evaluated at 20 $^{\circ}$ C is 0.32 day<sup>-1</sup>. Determine the critical oxygen deficit,  $D_c$ , critical time  $t_c$  and its location,  $x_c$ .

Data given:  $k_{1,T} = k_{1,20} \theta^{(T-20)}$ ;  $\theta = 1.056$ , and  $k_{2,T} = k_{2,20} \theta^{(T-20)}$ ;  $\theta = 1.047$  Saturation concentration of DO = 9.0 mg/l for temperature  $21^{\circ}$ C to  $22^{\circ}$ C

 $\mathbf{CO4} \quad (7)$