2nd Periodical Test (17 January 2022) Mass Transfer-1

Class- BE (Chemical & Food Tech.), 5th Semester Time Duration-1 hour Note Attempt all questions. M. Marks-25

Q.1 A packed tower is to be designed for the absorption of 98% of the ammonia from an airammonia mixture containing 4% ammonia at a rate of 4200 m³/h using water as the solvent. The tower operates at 105.1 kPa and 303 K. The equilibrium data for NH₃-water system at 303 K is given below:

Partial pressure of NH ₃	19.3	29.6	40.1	51.0	79.5	110
(mm Hg)						
Kg NH ₃ per 100 kg water	2	3	4	5	7.5	10

- (i) Calculate and plot the equilibrium data as x_A vs. p_A , x_A vs. y_A , y_A vs. y_A , y_A vs. y_A , y_A vs. y_A , y_A vs. $y_$
- (ii) Calculate the minimum liquid rate for the absorption (the inlet water is NH₃-free).

(5) (CO3)

Q 2 Explain the construction and working of sieve tray column.

(5) (CO4)

- Q 3 The air in a room is at 37.8 0 C and a total pressure of 101.3 kPa abs containing water vapor with a partial pressure is 3.59 kPa. Calculate (i) humidity. (ii) saturation humidity and percentage humidity, (iii) percentage relative humidity. (5) (CO4)
- Q 4 Write the various steps involved for the design of a cooling tower. (5) (CO4)
- Q 5 Develop a relation to find out time of batch drying. (5) (CO5)