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1<sup>st</sup> Mid-Term Exam

Class: B.E. Chemical 2<sup>nd</sup> Semester  
Subject: Material and Energy Balance  
Duration: 1 hour  
Total marks: 20  
Instruction:

- Answer all questions.
- Draw diagram for questions, wherever required.
- In case of conversions, write conversion factor.

1. Convert the following (8)
- 500nm to cm
  - 39.8 kg of NaCl per 100 kg water to kilogram moles of NaCl per kilogram mole of water
  - 0.4g/ (min m<sup>3</sup>) to lb<sub>m</sub>/ (hr ft<sup>3</sup>)
  - 1 horsepower to Watt.

2. Check dimensional consistency of following equation: (4)

$$q = 0.415(L - 0.2h_0)h_0^{1.5}\sqrt{2g}$$

Where q= volumetric flow rate (ft<sup>3</sup>/s)

L= crest height (ft)

h<sub>0</sub>= weir height (ft)

g= acceleration due to gravity (32.2 ft/s<sup>2</sup>)

3. The diagram shows a sketch of an artificial kidney, a medical device used to remove waste metabolites from the blood in case of kidney malfunction. The dialyzing fluid passes across a hollow membrane and the waste products diffuse from the blood into the dialyzing fluid. If the blood entering the unit flows at the rate of 220mL/min and the blood exiting the unit flows at the rate of 215mL/min, how much water and urea (the main waste products) pass into the dialysate if the entering concentration of urea is 2.3 mg/mL and the exit concentration of urea is 1.7mg/mL? If the dialyzing fluid flows into the unit at the rate of 1500 mL/min, what is the concentration of the urea in the dialysate? (8)

